

Ozone

Jacey Cressler // BS Public Health Student

EHST 3700: Industrial Hygiene Environmental Health Sciences Program Department of Health Education and Promotion East Carolina University Greenville, North Carolina

Properties and Occupational Uses

- Ozone (O₃) presents as a colorless-blue gas with pungent odor, condenses to a dark blue liquid or blue-black crystals.
- A chemical that is used in many industries.
- It can be used for purifying air and drinking water, decontamination, in industrial waste treatment, oil productions, bleaching and waxes, and even in order to make other chemicals.
- Ozone is both a natural and man-made chemical that occurs in the earth's upper and lower atmospheres.

Occupational Exposure

- Scenarios where workers are at risk of being exposed to ozone can include:
 - Factory workers in paper and pulp mills
 - Workers in wastewater treatment plants
 - Fisheries workers who treat storage water
 - Outdoor workers in areas with high levels of ozone

Toxicological Data

- The primary exposure of ozone occurs when workers breathe ambient air containing ozone in their occupational environment.
- Ozone has a pungent odor, a strong irritant, and highly toxic by inhalation. It is a strong oxidizing agent and a dangerous fire and explosion risk when in contact with organic materials.

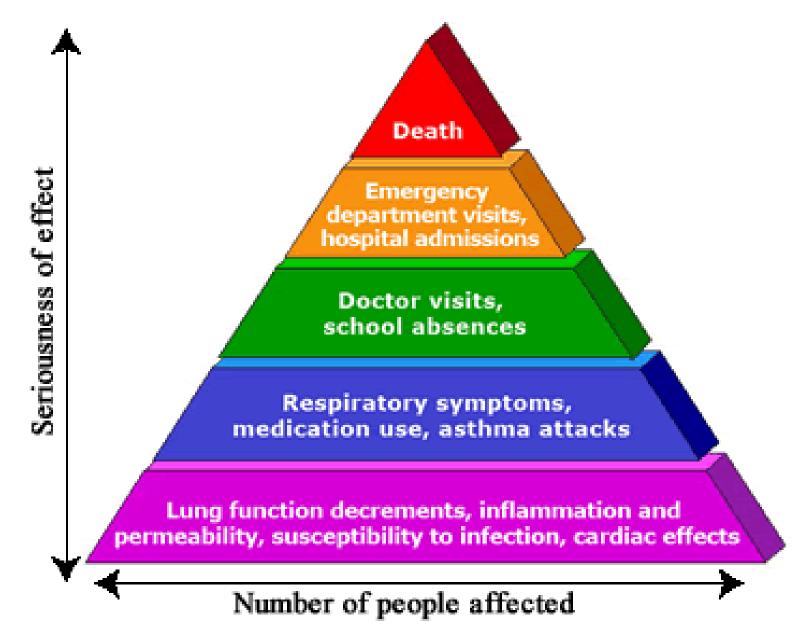


Figure 1. Pyramid of Effects Caused by Ozone

Epidemiological Studies

- Exposure to ozone may cause several adverse health affects; headaches, coughing, dry throat, shortness of breath, a heavy feeling in chest, and fluid in the lungs.
- Higher levels of exposures of ozone can lead to more severe respiratory symptoms. Chronic exposure may lead to asthma.
- The level of exposure depends upon the dose, duration, and work being done.

Sampling Methods

- The OSHA air sampling method for ozone consists of a calibrated sampling pump and a two-piece polystyrene cassette containing two nitrite-impregnated glass fiber filters.
- During monitoring, ozone will react with the nitrite on the filter collection device to convert to nitrate through oxidation. Using a flow rate of 0.5L/min with a sampling time of 180mins.

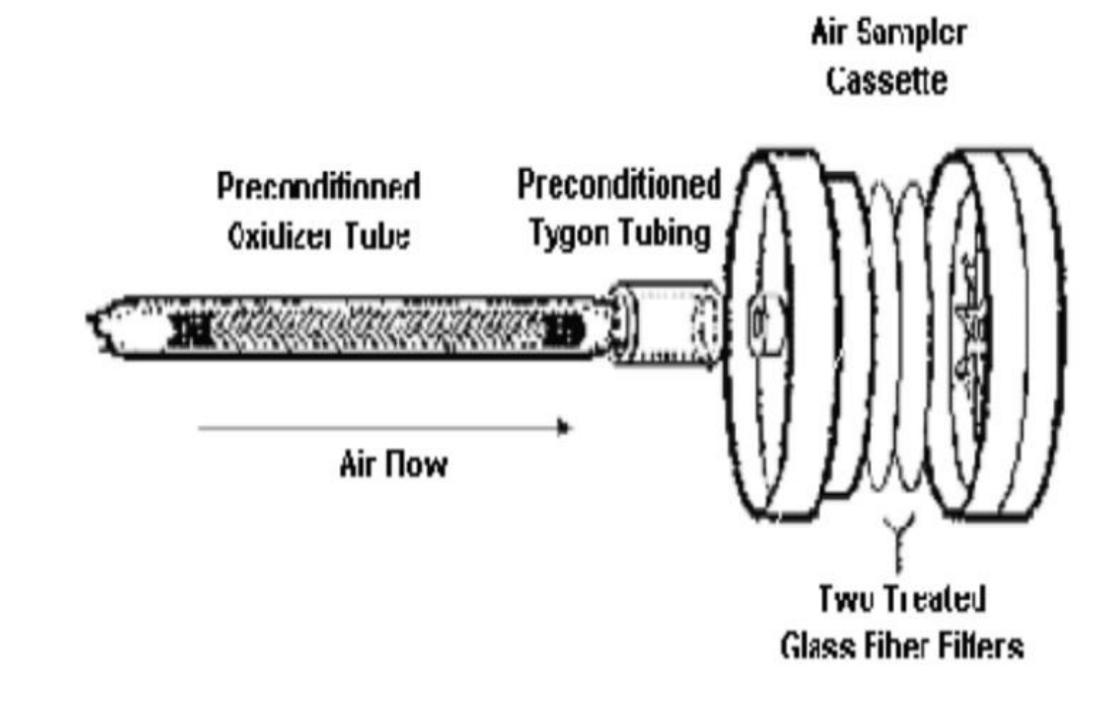


Figure 2. Ozone Sampler with Oxidizer Tube Source: https://www.osha.gov/sites/default/files/methods/osha-id214.pd f

Analytical Methods

- Ion Chromatography
- The reaction product then gets extracted from thew filters using deionized water to be analyzed with ion chromatography using a UV-VIS detector (200nm wavelength).
- Analyze samples, standards, and blanks according to Standard Operating Procedures.

Occupational Exposure Limits (OELs)

- The current OSHA permissible exposure limit (PEL) and NIOSH limit for ozone averaged over an 8-hour work shift is **0.1 part of ozone per million parts of air**.
- OSHA regulates employee exposure to ozone gas through its Air Contaminants Standard, 29 CFR 1910.1000.
- The ACGIH recommended airborne exposure limits are for heavy work, 0.05ppm; moderate work, 0.08ppm; light work, 0.1ppm; and workloads of less than 2 hours, 0.2pp, averaged over an 8-hour work shift.

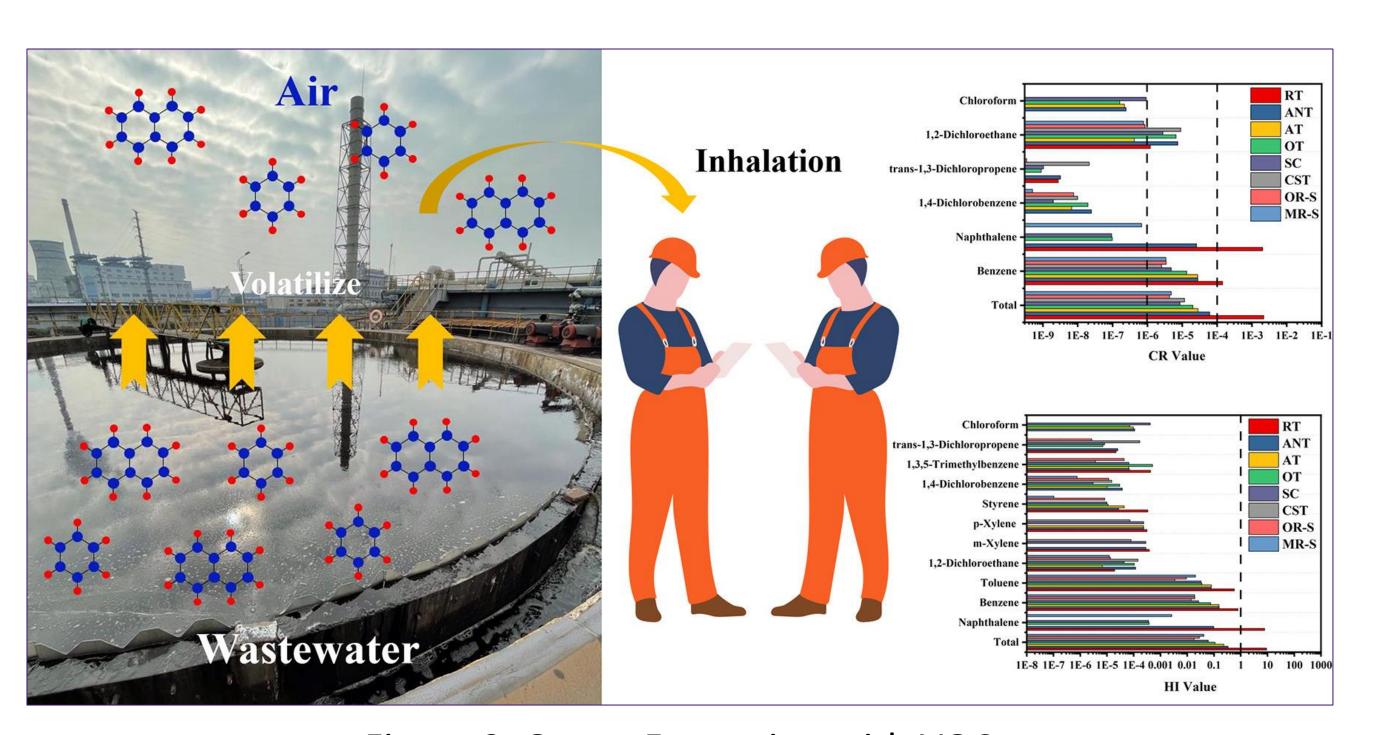


Figure 3. Ozone Formation with VOCs Source: https://ars.els-cdn.com/content/image/1-s2.0-S0048969722079487-ga1_lrg.jpg

Case Study

- Metal Inert Gas Welding and Tungsten Inert Gas Welding are widely used in the basic metal industries, which results in the usages of ozone and nitrogen oxides.
- A total of 84 welders (43 MIG/TIG welders, 41 non-MIG/TIG other) welders) were chosen, where concentrations of ozone and nitrogen oxide pollutants was measured in the breathing air of both workplaces.
- MIG/TIG Welders reporting pulmonary problems such as asthma, pulmonary inflammation, and higher susceptibility to infection.
- It seems that exposure of welders to ozone have a cumulative effect on their health.

Control Measures

- **Engineering Controls**: Elimination of ozone, if possible, increased local ventilation (not general), and ozone processes isolated/enclosed separately from work areas.
- Administrative Controls: Employee training on ozone as a hazardous substance, good work practices and maintenance to reduce exposures.
- Personal Protective Equipment: non-sorbent gloves and clothes, impact resistance goggles, full face respirator with gas cartridge approved for ozone.

References

<u>0exposure%20to%20ozone</u>.

- Centers for Disease Control and Prevention. (2019, June 22). Ozone. Centers for Disease Control and Prevention. Retrieved April 14, 2023, from https://www.cdc.gov/niosh/topics/ozone/default.html#:~:text=Exposure%20to%20ozone%20may%20cause,harmed%20from%2
- U.S. National Library of Medicine. (n.d.). Ozone. National Center for Biotechnology Information. PubChem Compound Database. Retrieved April 14, 2023, from https://pubchem.ncbi.nlm.nih.gov/compound/Ozone
- U.S. National Library of Medicine. (n.d.). Ozone. National Center for Biotechnology Information. PubChem Compound Database. Retrieved April 14, 2023, from https://pubchem.ncbi.nlm.nih.gov/compound/Ozone
- Ozone in workplace atmospheres (impregnated glass fiber filter) osha.gov. (n.d.). Retrieved April 14, 2023, from https://www.osha.gov/sites/default/files/methods/osha-id214.pdf
- Centers for Disease Control and Prevention. (2014, June 6). Occupational Health Guidelines for Chemical Hazards (81-123). Centers for Disease Control and Prevention. Retrieved April 14, 2023, from https://www.cdc.gov/niosh/docs/81-
- Ozone hazard summary ozone government of New Jersey. (n.d.). Retrieved April 14, 2023, from
- https://nj.gov/health/eoh/rtkweb/documents/fs/1451.pdf
- Azari, M. R., Esmaeilzadeh, M., Mehrabi, Y., & Salehpour, S. (2011). Monitoring of occupational exposure of mild steel welders to ozone and Nitrogen Oxides. Tanaffos. Retrieved April 14, 2023, from https://www.ncbi.nlm.nih.gov/pmc/articles/PMC4153171/